## CATALYZED INSERTION REACTIONS OF SUBSTITUTED lpha-DIAZOESTERS. A NEW SYNTHESIS OF CIS-ENOATES $^1$

Nobuo Ikota, Norio Takamura, Stanley D. Young and Bruce Ganem\*2

Department of Chemistry Baker Laboratories Cornell University Ithaca, New York 14853

Summary: The decomposition of  $\alpha$ -diazoesters is notably catalyst-dependent; with rhodium (II) carboxylates a stereoselective insertion reaction leads to the title structures in high yield.

The insertion of organic diazo compounds into acidic X-H bonds, catalyzed by a variety of acids<sup>3</sup> and transition metals,<sup>4</sup> has long constituted a powerful synthetic method for heteroatom alkylation.<sup>5</sup> Alpha-diazoesters have been used to effect difficult O-alkylations at neutral pH<sup>6a</sup> and to construct highly strained ring systems by NH insertion.<sup>6b</sup> The stereoselective formation of cis and trans-cinnamates from diazotized phenylalanine derivatives has also been reported.<sup>7</sup> Recently Pellicciari et al. described the Rh(OAc)<sub>2</sub>-induced conversion of  $\alpha$ -diazo- $\beta$ -hydroxyesters to  $\beta$ -ketoesters in high yield.<sup>8</sup> Our own interest since 1977 in the preparation of biologically important pyruvic acid enol ethers<sup>9</sup> led us to discover this same process while exploring the catalyzed insertion chemistry of various functionalized  $\alpha$ -diazoesters. Other new rearrangements we have noted now make possible some generally useful synthetic methods which are the subject of this Letter.

When methyl benzyloxy- $\alpha$ -diazopropionate  $\underline{1}^{10}$  was treated with methanol (1 equiv) and a trace of  $\mathrm{HBF_4(CH_2Cl_2,\ 0^\circ,\ 2h)}$ ,  $\mathrm{N_2}$  was evolved and acetal  $\underline{2}$  was produced in 54% yield [NMR & (CDCl\_3) 5.01 (t, 1H, J=6 Hz), 4.62 (m, 2H), 3.69, 3.37 (2s, 3 H each), 2.7 (d, 1H, J=6 Hz)]. In contrast,  $\underline{1}$  afforded exclusively cis- $\underline{3}$  upon exposure to a catalytic quantity of  $\mathrm{Rh(OAc)_2}$  in benzene [rt, 1h, 90%; NMR & (CDCl\_3) 6.41, 5.75 (AB quartet, 2H, J=6 Hz)]. We could not ascertain whether enol

ether  $\underline{3}$  was an intermediate in the HBF $_4$ -catalyzed formation of  $\underline{2}$ , or if the mixed acetal arose directly from  $\underline{1}$  by protonation and hydride migration. The rapid conversion of  $\underline{1}$  to a mixture of  $\underline{2}$  and  $\underline{3}$  (1:1, 80%) using Rh(OAc) $_2$  in CH $_3$ OH clearly demonstrated the susceptibility of  $\underline{3}$  to alcoholysis under very mild conditions. <sup>11</sup>

To test the hydride transfer mechanism, methyl benzoyloxy- $\alpha$ -diazopropionate  $\underline{4}$  was synthesized from N-Cbz-O-benzoylserine methyl ester. In CH<sub>3</sub>OH-CH<sub>2</sub>Cl<sub>2</sub> containing HBF<sub>4</sub>,  $\underline{4}$  slowly decomposed to afford a 4:1 mixture of  $\underline{5}$  and  $\underline{6}$  (36%). The major product was conclusively identified as  $\alpha$ -methoxy ester  $\underline{5}$  by DBU elimination (CHCl<sub>3</sub>, rt) to methyl  $\alpha$ -methoxy acrylate. This outcome was consistent with protonation of  $\underline{4}$  to an intermediate diazonium ion and partitioning between direct alcohol coupling and H-shift pathways. In contrast, treatment of  $\underline{4}$  with Rh(OAc)<sub>2</sub> (CDCl<sub>3</sub>, rt, 30 min) furnished only the rearranged  $\alpha$ -benzoyloxy acrylate  $\underline{7}$  [94%; NMR  $\delta$  (CDCl<sub>3</sub>) 6.10, 5.55 (AB quartet, 2H,  $\underline{J}$ =2Hz)].

$$\underline{5}$$
 R = H, R' = OCH<sub>3</sub>  
 $\underline{6}$  R = OCH<sub>3</sub>, R' = H

Metal-stabilized carbenoids possessing electrophilic character may be responsible for the unusual rhodium acetate-catalyzed reactions of  $\underline{1}$  and  $\underline{4}$ , and one plausible mechanistic picture is presented below.

$$\underbrace{\overset{P^{h}}{\overset{O}{\longrightarrow}}_{\operatorname{CO}_{2}\operatorname{CH}_{3}}}_{\underline{8}} \xrightarrow{CO_{2}\operatorname{CH}_{3}} \xrightarrow{2} \underbrace{\overset{T}{\longrightarrow}}_{H_{\overline{A}}} \xrightarrow{H_{\overline{B}}}_{RhL_{n}} \xrightarrow{2} \underbrace{\overset{CO_{2}\operatorname{CH}_{3}}{\overset{CH}{\nearrow}}_{H_{\overline{B}}}}_{\underline{9}} \xrightarrow{3}$$

In the case of  $\underline{4}$ , carbonyl participation through a 5-membered cyclic array as in  $\underline{8}$  (or its derived metal carbene) can lead to  $\underline{7}$  by benzoate migration. The fate of  $\underline{1}$  can also be rationalized by such a species  $\underline{9}$  wherein migration of  $H_A$  furnishes  $\underline{3}$  as a single isomer. This interpretation suggests a general synthetic access to cis-disubstituted enoates which has now been realized (see

Table). The decomposition of glutamate derivative  $\underline{12}$  was concentration dependent and generally afforded the desired cis-dimethylglutaconate [NMR & (CDCl<sub>3</sub>) 5.90, 6.65 ( $J_{vinyl}$  = 11 Hz)] in disappointing yield. Formation of dimer  $\underline{13}$  [mp 67-68°, m/e (CI) 345 (M<sup>+</sup>, 100%)] could be supressed by dilution, whereupon  $\underline{12}$  afforded predominantly dimethyl  $\alpha$ -hydroxyglutarate, perhaps reflecting  $\gamma$ -carbomethoxy participation as in  $\underline{8}$ . Besides the expected cis-vinylacrylate  $\underline{15}$ , diazopentenoic ester  $\underline{14}$  also furnished ethyl bicyclobutane-1-carboxylate  $\underline{16}$  in a remarkable reaction. Although strained ring structures undergo rearrangements catalyzed by various Rh and Pd complexes,  $\underline{16}$  proved stable towards Rh(OAc)<sub>2</sub> and Rh(OCOCF<sub>3</sub>)<sub>2</sub> for prolonged periods even at 80° in benzene. Isoleucine derivative  $\underline{17}$  decomposed essentially nonstereoselectively to an E-Z mixture of trisubstituted olefin  $\underline{18}$  in high yield.

The overall deamination of  $\alpha$ -aminoesters, which are readily accessible by straightforward alkylation routes, an efficient laboratory mimic of the biologically important ammonia lyase enzymes.

 $\frac{\text{TABLE}}{\text{Cis-Enoates from Rh(OAc)}_2\text{-Catalyzed $\alpha$-Diazoester Decomposition}^{\text{a}}}$ 

Diazoester		Product		Yield (Ref)
<u>1</u>		<u>3</u> .		90%
$(\operatorname{CH}_3)_2 \operatorname{CHCH}_2 \operatorname{C}(\operatorname{N}_2) \operatorname{CO}_2 \operatorname{CH}_3$	<u>10</u>	$^{\mathrm{cis-(CH}_{3})_{2}\mathrm{CHCH=CHCO}_{2}\mathrm{CH}_{3}}$		99% (16)
$PhCH_2C(N_2)CO_2CH_3$	11	cis-methyl cinnamate		100% (7)
$\mathrm{CH_3O_2CCH_2CH_2C(N_2)CO_2CH_3}$	12	cis-dimethyl glutaconate dimethyl $\alpha$ -hydroxyglutarate		14% 44% <sup>b</sup>
		$\begin{array}{c} \operatorname{CH_3O_2C} & \operatorname{CH_2CO_2CH_3} \\ & \operatorname{CO_2CH_3} \\ & \operatorname{CH_2CH_2CO_2CH_3} \\ & \underline{13} \end{array}$		62% <sup>C</sup>
$CH_2 = CHCH_2C(N_2)CO_2Et$	<u>14</u>	cis-CH <sub>2</sub> =CH-CH=CHCO <sub>2</sub> Et	<u>15</u>	39% (17)
		$ \overbrace{\mathrm{CO}_{2}^{\mathrm{Et}}} $ 16		51% (18)
Сң <sub>3</sub> сң <sub>2</sub> (сң <sub>3</sub> )снс(N <sub>2</sub> )со <sub>2</sub> сң <sub>3</sub>	<u>17</u>	CH <sub>3</sub> CH <sub>2</sub> (CH <sub>3</sub> )C≃CHCO <sub>2</sub> CH <sub>3</sub> E: Z 57:43	<u>18</u>	95% (19)

<sup>(</sup>a) To a solution of each diazoester in benzene (.1-.5M) was added a trace (1-2 mg) of

 $Rh(OAc)_2$  (Strem Chemical Co.). Immediate gas evolution was noted and after 30 min the solution was filtered through a short plug of  $30-60\mu$  silica gel using  $CHCl_3$ . Concentration at reduced pressure afforded the pure enoates whose IR and NMR matched the published spectra of authentic samples. (b) From a reaction at 0.019M; purified by flash chromatography. (c) From a reaction at 0.51M; recrystallized from ether-hexane.

Acknowledgment: Thanks are due to Messrs. R. Wood and R. Chinn for valuable experimental contributions. We are also grateful to the National Institutes of Health and the Petroleum Research Fund (administered by the American Chemical Society) for generous financial support.

## REFERENCES AND NOTES

- (1) Part 8 in the series "Shikimate Derived Metabolites." Part 7: Tetrahedron Lett., 715 (1979).
- (2) Fellow of the A. P. Sloan Foundation 1978-82; Camille and Henry Dreyfus Teacher-Scholar Grant Awardee, 1978-83.
- (3) (a) E. Muller, W. Rundel, <u>Angew. Chem.</u>, <u>70</u>, 105 (1958), (b) M. Caserio, J. D. Roberts, M. Neeman, W. S. Johnson, <u>J. Amer. Chem. Soc.</u>, <u>80</u>, 2584 (1958).
- (4) R. Paulissen, H. Reimlinger, E. Hayez, A. J. Hubert, Ph. Teyssie, Tetrahedron Lett., 2233 (1973).
- (5) For a review see W. Kirmse, "Carbene Chemistry," 2nd ed., Academic Press, New York, N.Y., 1971, especially p. 252ff.
- (6) (a) N. Ikota, B. Ganem, <u>J. C. S. Chem. Commun.</u>, 869 (1978); (b) T. N. Salzmann, R. W. Ratcliffe, B. G. Christensen, F. A. Bouffard, J. Amer. Chem. Soc., 102, 6161 (1980).
- (7) N. Takamura, T. Mizoguchi, S. Yamada, Tetrahedron Lett., 4267 (1973).
- (8) R. Pellicciari, R. Fringuelli, P. Ceccherelli, E. Sisani, J.C.S. Chem. Commun., 959 (1979).
- (9) B. Ganem, Tetrahedron, 34, 3353 (1978).
- (10) N. Takamura, T. Mizoguchi, K. Koga, S. Yamada, Tetrahedron, 31, 227 (1975).
- (11) We have additionally prepared  $\alpha$ -diazo- $\beta$ -hydroxypropionic esters <u>i</u> both by reducing dimethyl diazomalonate with DIBAL-H (hexane, -78°) and by condensing ethyl diazoacetate with paraformaldehyde (CH<sub>3</sub>OH, NaOH). Compound <u>i</u> rearranged quantitatively [catalytic Rh(OAc)<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, rt] to formylacetic ester <u>ii</u> thereby affording these water-soluble substances without resorting to a tedious extractive workup.

## $\underline{\mathbf{i}}$ HOCH<sub>2</sub>C(N<sub>2</sub>)CO<sub>2</sub>R OHCCH<sub>2</sub>CO<sub>2</sub>R $\underline{\mathbf{ii}}$

- (12) This perpendicular orientation of H<sub>A</sub> to the Rh-C bond is based on an earlier proposal concerning free carbene rearrangements: A. Nickon, F. Huang, R. Weglein, K. Matsuo, H. Yagi, J. Amer. Chem. Soc., 96, 5263 (1974).
- (13) K.C. Bishop III, Chem. Rev., 76, 461 (1976).
- (14) (a) G. Stork, A. Y. W. Leong, A. M. Touzin, <u>J. Org. Chem.</u>, <u>41</u>, 3491 (1976). (b) J. J. Fitt, H. W. Gschwend, <u>J. Org. Chem.</u>, <u>42</u>, 2639 (1977).
- (15) E. Havir, K. Hanson in "The Enzymes," 3rd Ed., P. Boyer, Ed., (New York, Academic Press, 1973), Vol. 7, p. 75.
- (16) D. V. Gardner, D. E. McGreen, Can. J. Chem., 48, 2110 (1970).
- (17) I. Moritani, Y. Yamamoto, H. Konishi, J. C. S. Chem. Commun., 1457 (1969).
- (18) K. B. Wiberg, R. P. Ciula, J. Amer. Chem. Soc., 81, 5261 (1959).
- (19) K. H. Dahm, B. M. Trost, H. Roller, ibid., 5294 (1967).

(Received in USA 26 June 1981)